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(54) Title: 3-(AMINO- OR AMINOALKYL)PYRIDINONE DERIVATIVES AND THEIR USE FOR THE TREATMENT OF HIV RELATED DISEASES

(57) Abstract

The present invention is concerned with 3-(amino- or aminoalkyl)pyridinone derivatives which inhibit the reverse transcriptase of the Human Immunodeficiency Virus (HIV). It relates moreover to the use of such compounds for treating HIV-related diseases. Furthermore it relates to a process for the preparation of these compounds.

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3-(Amino- or aminoalkyl)pyridinone derivatives and their use for the treatment of HIV related diseases

The present invention is concerned with 3-(amino- or aminoalkyl) pyridinone derivatives which inhibit the reverse transcriptase of the Human Immunodeficiency Virus (HIV).

It relates moreover to the use of such compounds for treating HIV-related diseases.

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Furthermore it relates to a process for the preparation of these compounds.

It is known that some pyrimidinone and pyridinone derivatives inhibit HIV reverse transcriptase.

In particular, derivatives from 1-[(2-hydroxyethoxy)methyl]-6-(phenylthio)thymine (HEPT) are well known for their HIV1 reverse transcriptase inhibitory properties.

European Patent Application EP-0 462 800 (Merck and Company Inc.) discloses pyridinones being substituted on position 3 with an aryl or heterocyclic group, linked to the pyridinone ring through a chain.

Unfortunately, strains resistant to these compounds appeared . Thus, their use in therapeutical treatments is questionable.

4-aryl-thio-pyridinones have been more recently disclosed by DOLLE et al. (1995, J. Med. Chem., 38, 4679-4686), and in the corresponding PCT Patent Application WO 97/05 113.

However, their activities are still moderate and their use in human therapy also could lead to the emergence of resistant strains.

The most active thio pyridinones disclosed therein have a 50% inhibitory concentration of virus multiplication (IC $_{50}$) for nevirapine resistant strains of about 260 nM.

The inventors have found a new pyridinone derivative family which show better HIV inhibitory properties.

They have moreover found a new process for obtaining these compounds.

The present invention relates to compounds having the following general formula I.

FORMULA (i)
$$R^{5} \xrightarrow{R^{6}} Q$$

$$X = R^{5}$$

wherein

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- Q represents -NR₁R₂ or -R₀NR₁R₂ wherein:
- *Ro represents C1-6 alkanediyl;
- * R_1 and R_2 each independently represent $C_{1\text{-}6}$ alkyl or $C_{3\text{-}6}$ alkenyl; said $C_{1\text{-}6}$ alkyl and $C_{3\text{-}6}$ alkenyl may be substituted with one, two or three substituents selected from hydroxy, $C_{1\text{-}4}$ alkyloxy, $C_{1\text{-}4}$ alkylthio, aryloxy, arylthio, amino, mono- or di($C_{1\text{-}4}$ alkyl)amino and aryl; or
- * R_1 and R_2 taken together may form a bivalent radical - R_1 - R_2 -wherein - R_1 - R_2 represents -(CH₂)₂-O-(CH₂)₂-, -(CH₂)₂-NR₇-(CH₂)₂,
- $-(CH_2)_2$ -CH(NHR₇)-(CH₂)₂- or -(CH₂)_n, wherein R₇ represents hydrogen or C₁₋₄alkyl and n represents 2, 3, 4, 5 or 6;
- R₃ represents aryl or a monocyclic or bicyclic heterocycle selected from pyridinyl, pyrimidinyl, thiazolinyl, furanyl, thienyl, imidazolyl, benzoxazolyl, benzothiazolyl, benzimidazolyl; said monocyclic or bicyclic heterocycle may optionally be substituted with one, two or three substituents each independently selected from hydroxy, C₁₋₄alkyl, C₁₋₄alkoxy, halo, trifluoromethyl, dimethylenoxy or phenyl,
- R_4 and R_5 each independently represent hydrogen, C_{1-6} alkyl, C_{3-6} alkenyl, C_{1-4} alkoxy, C_{1-4} alkyloxy, C_{1-4} alkyl, amino, mono- or di(C_{1-4} alkyl) amino, formyl, C_{1-4} alkylcarbonyl, carboxyl, C_{1-4} alkyloxycarbonyl, or C_{1-4} alkyl-

aminocarbonyl; wherein C_{1-6} alkyl and C_{3-6} alkenyl may be substituted with one, two or three substituents selected from hydroxy, C_{1-4} alkyloxy, C_{1-4} alkyl thio, aryloxy, arylthio, amino, mono- or di(C_{1-4} alkyl)amino and aryl; or

- R_4 and R_5 taken together form a bivalent radical of formula - R_4 - R_5 -wherein - R_4 - R_5 - represents -CH=CH-CH=CH- or -(CH₂)_t- , wherein t represents 3 or 4;

-R₆ represents hydrogen, hydroxy,C₁₋₄alkyloxy,C₁₋₆alkyl, C₃₋₆alkenyl, aryl, C₁₋₄alkyl, amino, mono- or di(C₁₋₄alkyl)amino or alkylaryl;

- Y represents O or S;
- X represents a radical of formula:

$$-(CH_2)_p$$
-
 $-(CH_2)_q$ - Z - $(CH_2)_r$ - or - CO-

wherein p represents 1, 2, 3, 4 or 5;

q represents 0, 1, 2, 3, 4 or 5;

r represents 0, 1, 2 or 3;

- Z represents NR8, C(= O), CHOH, CHNR8R9; CF2, O, S or CH=CH; wherein R8 and R9 each independently represent hydrogen or C1-4 alkyl;

or

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N-oxides, stereochemically isomeric forms or a pharmaceutically acceptable addition salts thereof.

As used in the foregoing definitions and hereinafter halo defines fluoro, chloro, bromo and iodo; C₁₋₄-alkyl defines straight and branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as methyl, ethyl, propyl, butyl and the like; C₁₋₆alkyl is meant to include C₁₋₄alkyl and the higher homologues thereof containing 5 to 6 carbon atoms such as, for example, pentyl, hexyl or the like; C₃₋₆alkenyl defines straight and branched chain hydrocarbon radicals containing one double bond and having from 3 to 6 carbon atoms, such as 2-propenyl, 3-butenyl, 2-butenyl, 2-pentenyl, 3-pentenyl, 3-methyl-2-butenyl and the like; and the carbon atom

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of said C_{3-6} alkenyl being connected to a nitrogen atom preferably is saturated; C_{1-6} alkanediyl defines bivalent straight and branched chain saturated hydrocarbon radicals having from 1 to 6 carbon atoms, such as, methylene, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl, 1,5-pentanediyl, 1,6-hexanediyl and the like. The term «C(=O)» refers to a carbonyl group. Aryl is phenyl or phenyl substituted with one, two or three substituents selected from C_{1-4} alkyl, C_{1-4} alkyloxy, halo and trifluoromethyl,

Preferred compounds according to the present invention are those in which X represents $-CH_2$ - or C (= O) and R₃ represents a phenyl group, substituted with two methyl groups, and the most preferred of them are those wherein R₃ represents a phenyl group substituted, in each meta position, with two methyl groups.

Preferably, in the compounds according to the present invention, R_1 and R_2 represent each a methyl group, R_4 represents an ethyl group, R_5 represents a methyl group and/or R_6 represents a hydrogen atom.

The most preferred compound of this invention is the 3-dimethylamino-4-(3,5-dimethylbenzyl)-5-ethyl-6-methylpyridin-2(1H)-one.

The compounds in which X is -CH₂-, R_3 represents a phenyl group optionally substituted, Y represents O and R_6 represents a hydrogen atom can be obtained by the general process represented on figure 1.

This first process comprises the following steps:

- a) reacting a pyridine (2), substituted in position 2 with an alkoxy group and in position 3 with an amidoalkyl group, with a C_1 - C_6 alkyllithium, resulting in a lithiated derivate (3) of the said pyridine.
- b) transforming the lithiated derivate (3) into an organocopper reagent by reacting it with a complex formed by Cu I and dimethyl sulphide.

- c) obtaining the pyridinone (4) by reacting the organocopper reagent with optionally substituted benzyl halide.
- d) hydrolysing the protected pyridinone (4) and obtaining the deprotected pyridinone (5).
- e) substituting the 3-amine group of the pyridinone (5) and obtaining the pyridinone (6).

This first process is summarized in the reaction Scheme I hereinafter:

SCHEME I

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In this process R_{10} and R_{11} represent independently C_1 - C_6 alkyl. In a preferred embodiment, R_{10} is a methyl group and R_{11} is a tert-butyl group.

The $C_1\text{--}C_6$ alkyllithium, reacted with the pyridine(2) can be a n-butyllithium.

The optionally substituted benzyl halide used in the step c) is preferably benzyl bromide.

The hydrolysis of the protected pyridinone(4), resulting in its deprotection, is advantageously obtained by adding hydrochloric acid to the pyridinone(4) and refluxing the mixture.

In a preferred embodiment, the amino group in position 3 of the pyridinone ring, deprotected during the step (d) is substituted by alkylation, by the Eschweiler-Clarke reaction.

Compounds wherein X represents $-(CH_2)_q$ -Z- $(CH_2)_r$ -, Y represents O, R₃ is an optionally substituted phenyl group and R₆ is an hydrogen atom can be obtained by a similar process.

Compounds wherein X represents C (= O), or -CH₂-, Y represents O, R_3 is an optionally substituted phenyl group and R_6 is an hydrogen atom can be obtained by a second process.

In this second process, the lithiated derivative (3) is reacted with an optionally substituted benzaldehyde, resulting in the intermediates of formula (7).

The intermediate (7) is oxidized to intermediate (8).

The intermediate (8) is thereafter deprotected by hydrolysis, as in the first process, resulting in the pyridinone (9) of general formula I.

This second process is summarized in the reaction scheme II hereinafter.

Reaction scheme II

Preferably the oxidation of the intermediate (7) is performed in the presence of manganese dioxide.

The intermediate (7) can also be transformed into corresponding ester (10) wherein R_{12} represents a C_1 - C_4 alkyl group whose hydrogenolysis provides pyridinone(4) in better yields. Preferably, the ester (10) wherein R_{12} is CH_3 is prepared by treatment of intermediate (7) with acetic anhydride. Subsequently hydrogenolysis is performed under hydrogen atmosphere and in the presence of a catalyst, especially 30% paladized charcoal. This process is summarized in the reaction scheme III

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Reaction scheme III

$$R10$$
 $R10$
 $R10$

Other compounds of general formula I, and wherein X is $(CH_2)_p$ or $(CH_2)_q$ -Z- $(CH_2)_r$ or C(=O), and R_3 is other than phenyl and R_6 is other than hydrogen can be obtained by these processes, appropriately adapted by the man skilled in the art.

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The compounds according to the present invention, in which X is S can be obtained by the process described in the article of **DOLLE et al.** (1995, previously cited) or in the corresponding patent application WO 97/05 113, the contents of which are included in the present application.

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The compounds can also be obtained by other processes known by the man skilled in the art.

The present invention relates moreover to the intermediates of the processes hereabove disclosed. In particular it relates to the lithiated derivative of formula (3).

The compounds of the present invention are useful in the inhibition of HIV reverse transcriptase, and in particular HIV-1 reverse transcriptase and the prevention or treatment of infection by the human immuno deficiency virus (HIV) and of HIV-related diseases, such as AIDS.

For these purposes, the compounds of the present invention may be administered orally, parenterally (including sub-cutaneous injections, intravenous, intramuscular, intrasternal injection or infusion tectoniques), by inhalation spray, or rectally, in dosage unit formulations containing pharmaceutically acceptable carriers, adjuvants and vehicles.

Thus, another object of the present invention is a method, and a pharmaceutical composition for treating HIV related diseases, HIV infection, and in particular AIDS.

The invention relates also to these compounds for use as medecine and to their use for the manufacture of a medecine for the treatment of HIV related diseases, HIV infection, and in particular AIDS.

These pharmaceutical compositions may be in the form of orallyadministrable suspensions or tablets, nasal sprays, sterile injectable preparations, or suppositories.

The present invention is illustrated without being limited by the following examples.

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EXAMPLES:

EXAMPLE 1

<u>Preparation of 3-dimethylamino-4-(3,5-dimethylbenzyl)-5-ethyl-6-methylpyridin-2(1H)-one.</u>

1) 5-Ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine.

This compound has been prepared as indicated by **DOLLE et al**. (1997, Tetrahedron, vol.53, n°37, 12.505-12.524). The content of this article is hereby incorporated by reference.

3,68g of 3-Amino-5-ethyl-2-methoxy-6-methylpyridine (22,14 mmol), obtained as indicated by HOFFMAN et al. (1993, J. Med. Chem., 36, 953-966), was dissolved in a mixture of dichloromethane (260 ml) and triethylamine (3.39 ml). The mixture was cooled at 0°C and 3.00 ml of trimethylacetyl chloride was added dropwise. The solution was stirred at 0°C for 15 min. and then washed with 100 ml water. The aqueous layer was extracted with 3 x 200 ml dichloromethane. The combined organic layers were dried over magnesium sulfate and concentrated under reduced pressure. The residue was purified by column chromatography using dichloromethane as eluant to provide the 5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine (5.31g; 96%). Elemental analysis calculated for C₁₄H₂₂N₂O₂; C, 67.17. H, 8.86; N, 11:19; O, 12.78; found : C, 67.11; H, 8.56; N, 10.91; 0, 12.67.

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2)4-(3,5-Dimethylbenzyl)-5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine

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i) By lithiation of 5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine:

5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine and 3,5-dimethylbenzyl bromide were dried in the presence of phosphorus pentoxide under vacuum at room temperature during 24 hours. Copper iodide (Cu^II) was dried in the presence of phosphorus pentoxide under vacuum at 50°C for 24 hours. 5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine (1.06g) and freshly distilled tetramethylethylenediamine (TMEDA) (2.24 mL) were dissolved in dry tetrahydrofuran (THF) (26 mL) and the mixture was cooled at -78°C under a nitrogen atmosphere. n-Butyllithium (1.6 M in hexane, 9.26 mL) was added dropwise. The mixture was stirred for 1 hour at O°C.

Cull :dimethyl sulfide complex, prepared by adding dimethylsulfide (14 mL) to a suspension of copper iodide (2.82g) in dry THF (52 ml) at -78°C under N2 atmosphere, was then added dropwise to the mixture at -78°C. The mixture was stirred at O°C for 30 min and cooled again at -78°C to allow the addition of 3,5-dimethylbenzyl bromide (3.81g) dissolved in THF (4 mL). The resulting mixture was stirred at O°C for 3 hours and at room temperature for 12 hours. 16 mL of water and 20 mL of 28% aqueous ammonium hydroxide were added. The aqueous layer was extracted with 3 x 80 mL of ether. The combined organic layers were washed with 40 mL of brine, dried over magnesium sulfate and concentrated under reduced pressure. The residue was purified by column chromatography using 4-(3,5cyclohexane-ethyl acetate (1:0 to 8:2) as eluant giving (577 dimethylbenzyl)-5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine mg, 37%) mp 138-139°C.

ii) By hydrogenolysis of ± (5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridin-4-yl)-(3,5-dimethylphenyl)-methyl acetate.

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(+, -) (5-Ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridin-4-yl)-(3,5-dimethylphenyl)-methylacetate.

8.34g of (+, -)-(3,5-dimethylphenyl)-(5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridin-4-yl)-methanol, prepared as described below, was dissolved in pyridine (200 mL) and added to acetic anhydride (10.24 mL), and the solution was stirred for 1.5 h at room temperature and for 60 h at 60°C. An additional 10.24 mL of acetic anhydride (108.51 mmol) was added and heating was continued at 60°C for 24 h. The pyridine was evaporated under reduced pressure and the residue was taken up in 500 mL of ethyl acetate. The organic layer was washed with 170 mL of an aqueous saturated sodium bicarbonate solution, 170 mL of water and 170 mL of brine, dried over magnesium sulfate and the solvent was evaporated. The residue was purified by column chromatography using dichloromethane-ethanol (1:0 to 95:5) to give the titled compound (8.78g, 95%) mp 70-71°C.

A mixture of this compound (850 mg) and Pd-C (30%, 850mg) in acetic acid-water-dioxane (42.5 mL, 2:1:2, v/v/v) was stirred at room temperature for 24 hours under 10 atm of hydrogen. The catalyst was removed by filtration and washed with ethanol. The solvent of the combined filtrates was evaporated under reduced pressure giving 4-(3,5-dimethylbenzyl)-5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine (726 mg, 99%) which was identical to the compound as prepared in example 1.2.i).

3) 3-Amino-4-(3,5-dimethylbenzyl)-5-ethyl-6-methylpyridin-2(1H)-one.

3M aqueous hydrochloric acid (150 mL) was added to a suspension of 4-(3,5-dimethylbenzyl)-5-ethyl-2-methoxy-6-methyl-3-pivaloylaminopyridine (2.36 g) in water (300 mL). The mixture was refluxed

for 3.5 h and then stirred at room temperature for 12 h. The solution was basified by adding concentrated ammonium hydroxyde and was extracted with 3 x 800 mL ethyl acetate. The combined organic layers were washed with 110 mL brine, dried over magnesium sulfate and concentrated under reduced pressure giving 3-amino-4-(3,5-dimethylbenzyl)-5-ethyl-6-methylpyridin-2(1H)-one . (1.79g, 100%). mp 204-205°C.

4) 3-Dimethylamino-4-(3,5-dimethylbenzyl)-5-ethyl-6-methylpyridin-2-(1H)-one.

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To a stirred solution of 3-amino-4-(3,5-dimethylbenzyl)-5-ethyl-6-methylpyridin-2(1H)-one (200 mg) and 37% of aqueous formaldehyde (0.60 mL) in 5 mL of acetonitrile was added 139 mg of sodium cyanoborohydride. Glacial acetic acid (0.07 mL) was added dropwise and the reaction mixture was stirred at room temperature for 2 hours. An additional 0.07 mL of glacial acetic acid was added, and stirring was continued for 30 minutes. The solvent was evaporated and 15 mL ether were added to the resulting residue. The organic layer was washed with 3 x 30 mL 1N aqueous potassium hydroxide and 3 mL brine, dried over magnesium sulfate and concentrated under reduced pressure to give 3-dimethylamino-4-(3,5-dimethylbenzyl)-5-ethyl-6-methylpyridin-2(1H)-one (200 mg, 91%) mp 229-230°C.

EXAMPLE 2: 1) Biological activity of the compound according to example 1.

1. Material and Methods

The antiviral activity, the expression and purification of the recombinant HIV-RT enzyme, the reverse transcriptase activities and the inhibition of RT were evaluated as described in WO 97/05 113.

The retrovirucidal effect and the reverse transcription were measured as described hereinafter.

1.1. Retrovirucidal effect.

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HIV-1 viral suspensions were obtained by coculture of MT4 cells and H9 cells chronically infected by HIV-I_{Lai} isolate. 200µl of a cell supernatant containing viral particles (HIV-ILai: 100 TCID50) were incubated at room temperature with various concentrations of different inhibitors. After 3 hours, virions were washed through 0.02µm anopore membrane in 1.5 mL Vectaspin tube (Whatman) for 10 minutes at 5 000 g . Each of the three subsequent washes was performed in the same conditions after the viral concentrate was refilled with 500 µL of RPMI medium. Then, the viral concentrate was readjusted to the initial volume with RPMI plus 10% foetal calf serum (FCS). The residual infectivity was assayed on P4 cells as described by CHARNEAU et al.. (1994, J. Mol. Biol., 241, 651-662). Briefly, P4 cells were plated using 100 µL of DMEM medium plus 10% FCS in 96 plate multi-wells at 20 x 10⁵ cells per mL. After overnight incubation at 37°C, the supernatant was discarded and the viral preparation (200 µL) was added. One day later the wells were washed three times in PBS. Each well was refilled with 200 µL of a reaction buffer containing 50 mM Tris-HCl pH 8.5, 100 mM 2-mercaptoethanol, 0.05% Triton X-100 and 5 mM 4methylumbelliferyl β-D-galactopyranoside (MUG). After 3 hours at 37°C, the level of the reaction was measured in a fluorescence microplate reader.

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1.2) Reverse transcription.

The plasmid pAV4 containing the 50-997 HIV-1 nucleotide fragment (MAL strain) in pSP64, under the control of the bacteriophage T7 promoter was a kind gift from Dr. J.L. DARLIX (INSERM-Lyon, France). E. coli HB 101 recA was used for plasmid amplification. After digestion of this clone with Pstl and in vitro transcription using T7 RNA polymerase, a HIV-1 genomic RNA fragment starting at position +50 of the MAL sequence was obtained. In vitro transcription using T7 RNA polymerase as performed as follows. Three μg of linearized plasmid DNA were transcribed in 100 μL of 40 mM Tris -HCl pH 8.O, 8 mM MgCl₂, 10 mM spermidine, 25 mM NaCl, 10 mM dithiothreitol, 0.5 mM of each ribonucleoside triphosphate, with 100 units of T7 RNA polymerase and in the presence of 20 units of human placenta ribonuclease inhibitor, for 2 hours at 37°C. After treatment with 12 units of Rnase-free Dnase I (for 10 minutes at 37°C), the RNA transcripts were extracted with 1 volume of phenol/chloroform/isoamyl alcohol (24:24:I) and with chloroform and precipitated in 2.5 volumes of ethanol and 0.3 M ammonium acetate (pH 5.5).

Reverse transcription was performed in a total volume of 50 μ L containing 50 mM Tris-HCl pH 8.0, 6 mM MgCl₂, 2 mM dithiothreitol, 12 mM NaCl, 150 nM HIV-1 RNA, and either 200 nM of a synthetic oligodeoxynucleotide primer (18-mer ODN) complementary to the PBS of HIV-1 RNA, or 200 nM tRNA^{Lys3}. When the 18-mer ODN was used as primer, incubation was carried out at 37°C with the template and 300 nM RT. After 30 minutes, 10 μ Ci [α -³²P]dGTP (3000 Ci/mmol) and 0.1 mM of each dNTP were added and the incubation proceeded for 30 minutes at 37°C. With tRNA^{Lys3} as primer, the same conditions were used except that tRNA and RNA were prehybridized by heating for 2 minutes at 90°C and then slowly cooled. Samples were extracted with phenol-chloroform and

collected by ethanol precipitation. Reaction products were analyzed on 8% polyacrylamide-TBE (90 mM Tris pH 8.3, 90 mM borate, 2 mM EDTA)-7 M urea gels.

5 RESULTS

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The antiviral activity of the compounds according to example 1 has been tested on various strains.

On HIV-LAI wild type this compound shows the following activities: $10 - 1050 = 0.2 \, \text{nM}$; CC50 > $10^5 \, \text{nM}$ (S.I. > 33.333).

On an HIV-1 novirapine resistant strain the activities of the compound of example 1 are as follows:

 $IC_{50} > 10^4 nM$

CC₅₀ > 10⁴nM

The compound of example 1 has been also tested on various HIV strains and primary cell cultures. The table 1 illustrates the activity of this compound on these strains.

The retrovirucidal effect of the compound according to example 1 has been tested. Table 2 illustrates this effect at various doses of this compound.

The IC_{50} of the compound of example 1 for the inhibition of the reverse transcriptase is 20 nM.

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TABLE 1-Anti HIV-1 activity of the compound of example 1 on various HIV strains and primary cell cultures IC₅₀(nM)/CC₅₀(nM)

Compound	HIV-1 IIIIB	HIV-1	HIV-1 IIIB	HIV-2 D	HIV-1 Bal/
	/MT4	AZTres.	/PBMC	194	Mono/macro-
		/MT4		/PBMC	phages
Example 1	2.4/>1000	0.2/>1000	0.58/>1000	>1000/>	0.004/>1000
				1000	

TABLE 2: Inhibition of infectivity of the compound of example 1

Dosage of compound of example 1	% inhibition of infectivity
10 nM	26%
100 nM	46%
1 μm	83%
10 μm	99%

EXAMPLE 3: Other 3-(amino- or aminoalkyl) pyridinone derivatives and their retrovirucidal activity against two different HIV-1 strains.

3.1 Compounds:

Further compounds according to the general formula (I) (compounds n°1-25, 27-108, 110-125, 127-145 and 147-203) as well as four intermediate compounds used for synthesis (compounds n°26, 109, 126 and 146) have been synthesized and are listed in table 3 below.

The meaning of each of the groups Y, Q and R3 - R6 is defined for every exemplified pyridinone derivative.

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3.2 RETROVIRUCIDAL EFFECT

The retrovirucidal effect of each pyridinone derivative listed in table
3 has been assayed according to the teachings of example 2, excepted that
the anti-viral effect has been tested on the two following HIV-1 strains:

18

- a) HIV-1 strain IIIB (see example 2);
- b) HIV-1 strain 103 N which is a mutant strain bearing a point
 mutation in the reverse transcriptase gene leading to an enzyme wherein the initial Lys-103 residue is replaced for a Asn residue.

HIV-1 103N strain exhibits resistance to the reverse transcriptase inhibitor TIBO R82913 (BALZARINI J. et al. 1993, Virology, 192: 246-253). The HIV-1 103 N strain has also been described by SAHLBERG et al.,(1998, Antiviral Res., 37 (3): ASS) and BALZARINI et al. (1996, Antimicrobial Agents and Chemotherapy, 40 (6): 1454-1466).

The results are expressed as pIC_{50} ($pIC_{50} = -\log IC_{50}$), of every of compound as regards to each of the HIV-1 strains IIIB and 103N. Thus, the pIC_{50} value of compound n°1 as regards to HIV-1 IIIB being 7,6999, the IC_{50} can be directly deduced as being equal to $10^{-7,6999}$ M.

Such high retrovirucidal activities had never been observed previously when using prior art reverse transcriptase inhibitors.

Consequently, the novel pyridinone derivatives according to the present invention are of a high therapeutical value against HIV related diseases, particularly against HIV-1 related diseases.

TABLE 3

							HIVI	pIC50
	Υ	Q	R3	R4	R 5	R6	strain IIIB	strain 103N
1	0	NH2	Chemistry 4	Et	Ме	н	7.699	6.671
2	0	NH2	3,5-Dimethylbenzoyl	Et	Ме	н	6.612	6.64
3	0	NMe2	3,5-Dimethylbenzyl	Et	Ме	н	8.004	7.438
4	0	Chemistry 33	3,5-Dimethylbenzyl	Et	Me	н	5.094	<4
5	0	NH2	3,5-Dimethylbenzyl	Et	Me	Н	6.261	5.636
6	0	NH2	Chemistry 52	Et	Me	н	5.795	5.026
7	0	NH2	Chemistry 58	Et	Me	н	<4	~4
			Ò					4.39
	0	NH2 	4-Methylbenzyl	Et	Ме	н	5.373	5.103
, ,	0	NH2	3-Methylbenzyl	Et	Me	Н	1 5.3/3	3.103

10	0	NMa2	Chemistry 82	Et	Ме	н	6.241	4.389
11		NMe2	3.5-Dimethylbanzyl	Eı	Ме	Ме	7.215	6.094
12	0	NEt2	3,5-Dimethy/benzyl	Et	Ме	н	8.022	6.363
13	o	NMe2	3-Methylbenzyl	Et	Ме	н	8.824	7.622
14	0	NMe2	2-Methylbenzyl	Et	Ме	Σ.	7.676	5.849
15	0	NH2	3.5-Dimethy/benzyl	н	н	н	<4.17	4.138
16	0	NM62	3,5-Dimethylbenzyl	н	н	н	5.061	4.401
17	0	N(n-Pr)2	3.5-Dimethy/benzyl	Et	Me	н	6.285	4.379
18	30	NMe2	4-Methylbenzyl	Et	Me	н	6.454	4.895
			4					5,947
	00	NMe2	3,4-Dimethylbenzyl	Et Et	Me Me	H	6.926	5.585

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			1				,
21 0	NMe2	Benzyl	Ει	Me	н	8.409	6.65
		ኣ					
1 1							
220	NMe2	3,5-Dimethylbenzyl	Et	Me	Benzyl	4.603	<4
	in the second se	×					
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					0		
23 0	NMe2	3.5-Dimethylbenzyl	El	Me	Chemistry 163	5.254	<4
	\ \\]					
	x N						
24 0	Chemistry 165	3,5-Dimethy/benzyl	Et	Ме	н	4.262	<4
		\					
25 o	Chemistry 171	3,5-Dimethylbenzyl	Eı	Me	н	<4	4.259
		× 40					
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	o L						
26 o	Chemistry 177	3,5-Dimethylbenzoyl	Et	Me	н		
		ነ ጓ					
27 o	NH2	3,5-Dimethylbenzyl	Me	Et	н	5.949	5.098
		\					
28 o	NMe2	3,5-Dimethylbenzyl	Me	Et	Н	8.032	6.943
	10002	×		T			
				1	}		
29 o	NHCH2Ph	3,5-Dimethylbenzyl	Et	Мө	н	6.555	5.496
		1 1					
	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\						
30 o	Piperidin-1-yl	3,5-Dimethylbenzyl	Et	Ме	н	6.214	4.224
		1 4					
31 0	NH2	2,4-Dimethylbenzyl	Et	Me	н	<4	<4

32 0 NH2 3,5-Dimethylbenzyl Me Me H 6.104	<5
	<5
33 0 NMe2 3.5-Dimethylbenzyl Me Me H 8.42	
33 O NMe2 3,5-Dimethylbenzyl Me Me H 8.42	
X	6.286
1	
	_
34 O NMe2 2.4-Dimethylbenzyl Et Me H 5.019	<4
35 O NMe2 3.5-Dimethy/benzoyl Et Me H 8.585	7.987
	į
36 O N-Morpholino 3,5-Dimethylbenzyl Et Me H 6.763	<4
37 O NMe2 2,5-Dimethylbenzyl Et Me H 6.796	5.729
38 0 NMe2 3,5-Dilluorobenzyl Et Me H 8.155	7.402
39 0 NH2 3-Chlorobenzyl EI Me H 5	4.751
40 O NMe2 3-Chlorobenzyl Et Me H 8.585	7.412
41 O NH2 3-Fluorobenzyl Et Me H 5.131	4.473
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			1	,			
		1					
							6.400
43 o	NMe2	Chemistry 280	Et	Me	Н	7.377	6.422
				1			
44 o	NMe2	Chemistry 286	Et	Ме	н	7.889	6.355
440	NMEZ	*					
45 o	NMe2	3,5-Dimethylbenzyl	Et	Me	Et	5.519	4.095
		ኝ					
46 o	NHMe	3,5-Dimethylbenzyl	Et	Me	Н	8.119	7.034
		1 1					
	*N~~						
47 0	Chemistry 303	3,5-Dimethylbenzyl	Et	Me	<u> </u> н	7.767	6.968
							6.711
480	NMe2	Chemistry 310	Et	Me	Н	8	6.711
			i				
49 0	NH2	Chemistry 316	Et	Me	н	<4	<5
		1 2					
50	NH2	3-Trifluoromethylbenzyl	Eı	Me	н	<5	< 5
		*					
							_
51	D NMe2	Chemistry 334	Et	Me	H	5.384	<5
		F F	Eı	Me		<4	<5
52	0 NH2	4-Trilluoromethylbenzyl	- Ier	Ive]H		

			_	,			·····
53 o	NMe2	4-Trifluoromethylbenzyl	Et	Me	н	5.828	<5
54 o	NH2	4-Chlorobenzyl	Et	Me	н	<4	< 5
		Ò				6.851	
55 o	NM62	3,5-Dimethylbenzyl	Et Et	Me Me	н	6.651 8.194	7.11
	Chemisty 303	J.					
57 o	NMe2	3-Trifluoromethylbenzyl	Et	Me	H	8.086	6.414
58 o	NH2	2.4.6-Trimethylbenzyl	E1	Me	Н	<4	< 5
59 o	NMe2	2,4,6-Trimethylbenzyl	Et	Ме	н	5.029	<5
60 o	NMe2	3.5-Dimethylbenzyl	Et Et	Me Me	H	7.693	7.001 5.922
62 0	Chemistry 393	3,5-Dimethybenzyi	Et	Me	н	6.604	5.305

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63	_	NMe2	3.5-Dimethylbenzyl	Me	n-Pr	н	7.029	6.334
03	<u> </u>	NM82	3.5 Onneury Menzyr			~	7.023	
	1]					-
64	0	NHC(=0)-iPr	3,5-Dimethylbenzyl	Et	Me	н		
1			ኣ					
				- .		<u>.</u>	8.284	6.405
65	<u> </u>	NMe2	2-Chlorobenzyt	Et	Me	н	6.284	6.405
			<u> </u>				-	
			N S					
66	0	NMe2	Chemistry 430	Et	Me	н	7.588	5.72
			ጓ					
1								
67	_	A V V	3,5-Dimethy/benzyl	Et :	Me	н	6.804	4.955
67	-		3,5-Dittedsylberizyr		1410	<u>'</u>	0.007	
		Y H	1					i i
		,						1
68	0	Chemistry 441	3.5-Dimethylbenzyl	Et	Мв	н		
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				1				
	ĺ			_		l I	6 901	5.655
69	0	NH(n-Bu)	3,5-Dimethylbenzyl	EI	Me	н	6.891	5.033
			Ì	1 7				
				$ \ \ \ $				
70	0	NMe2	3,5-Dimethylbenzyl	Chemistry 45	Ме	н	7.752	7.159
			×					
1							ļ	
71	0	NMe2	3,5-Dimethylbenzyl	n-Pr	Me	н	7.777	7.049
				İ				
		x [™] ✓						
72	20	Chemistry 465	3,5-Dimethylbenzyl	Et	Me	н	7.079	<4
<u> </u>	1		***	1				
73	3 0	NH2	Chemistry 472	Et	Ме	н	8.027	6.92

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74 o	NH2	Chemistry 478	Et	Мв	н	<4	<4
		ኣ					
			ŀ				
75		N 100	_	Me	ļ h	5.252	4.132
75 o	NMe2	Chemistry 490	Et	Me		3.232	4.102
	1				1		
76 o	NH2	3,5-Dimethylbenzyl	н	i-Am	H	<5.494	<4
		1 1					
			1				
77 0	NMe2	3,5-Dimethylbenzyl	н	i-Am	н	5.827	<4
		٦					
	x"~~~~						
78 0	Chemistry 507	3,5-Dimethylbenzyl	Et	Me	н	8.678	7.128
		ኣ			1		

79 0	Chemistry 513	3,5-Dimethylbenzyl	Et	Ме	н	6.987	5.47
130	Offernatily 313	× ^					
		HN J					
800	NH2	Chemistry 520	Et	Me	H	<4	<4
			1.			İ	
81 0	NHEt	3,5-Dimethylbenzyl	Et	Ме	н	7.866	6.444
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	x#~~						
82 0	Chemistry 531	3,5-Dimethylbenzyi	Et	Me	н	7.735	5.813
		× 🗥					
		HN					
83 0	NH2	Chemistry 538	Et	Me	H	<4.033	<4
84 0	NH2	Chemistry 544	Et	Me	н	<4	<4
		1 1					
85	NH2	3-Methylbenzyl	Me	Ме	н	4.954	<4

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86 0		N11.4-0	3-Methylbenzyl	Me	Мв	н	7.863	5.936
80 0		NMe2		Me	MB		7.000	0.550
			\ Y°					
87 0		NII 40	3-Methylbenzoyl	Et	Me	н	6.46	5.653
8710	,	NH2	3-Metryloenzoyi	-				
				1			l	
88)	NMe2	Chemistry 568	Et	Ме	н	<4	
			ኣ					
				<u>l</u>		l l	6.237	
89		NH2	3,5-Dimethylbenzyl	н	n-Bu	H	6.237	
							İ	
90	5	NMe2	3,5-Dimethylbenzyl	н	n-Bu	н	6.359	
			×					
				Į				
91	0	NH2	3-Methylbenzyl	(CH2)4	(CH2)4	н	5.73	
			ጓ		•			
	_			(CH2)4	(6) 10) 4	 H	7.807	
92	0	NMe2	3-Methylbenzyl	(GHZ)4	(CH2)4		7,007	
1 1			\ *\^\\					
				1				
93	^	NMe2	3-Methylbenzoyl	Et	Me	н	8.721	•
		111102	V .0					
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94	0	NH2	3-Methylbenzoyl	Me	Me	н	5.153	
			¥,0°					
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				1				
95	0	NE ₁₂	3-Methylbenzoyl	Et	Ме	н	8.268	
			×p°	1		1 1		
96	0	NMe2	3-Methylbenzoyl	Ме	Me	н	7.824	6.37
			1 3 6					
	.[) N		1.	1		
97	10	NH2	Chemistry 622	Et	Ме	Н	<4	<4

			à					
98	0	NH3	3-Ethylbenzyl	Et	Me I	н	5.358	4.978
99	0	NMe2	3-Ethy/benzyl	Ει	Me	н	8.569	6.718
100						u	4.871	<4
100	0	NH2	3,5-Dimethy/benzyl	Н	Me	н	4.071	
101	o	NMe2	3,5-Dimethylbenzyl	н	Me	н	6.341	4.25
102	0	NMe2	Chemistry 652	Et	Me	н	4.369	<4
103	o	NH2	Chemistry 658	Et	Ме	н	5.747	
104	0	NMe2	Chemistry 684	Et	Ме	н	8	7.058
105	50	NH2	3.5-Dimethylbenzyl	СІ	н	н	4.943	
100	6 o	NM02	3.5-Dimethylbenzyl	CI	н	н	7.063	
10	7 0	NMe2	3-Methylbenzoyl	(CH2)4	(CH2)4	н	7.231	
	80	NM62	3-Methylbenzoyl	Ма	Et	н	7.005	
	90	Chemistry 699	3.5-Dimethylbenzyl	н	OMe	н		

								
110			он о				7.783	
110	0	NMe2		Et	Me	н	7.703	
111	0	NH2	Chemistry 712	Et	Me	н	< 4	
			7 .0					
112	o	NMe2	Chemistry 718	Et	Me	н	6.394	
	ļ		×					
113	0	NH2	Chemistry 724	Et	Me	н	5.273	
		r ii						
114		Chemistry 729	Chemistry 730	Et	Ме	н		
115			Š	Et	Me	Chemistry 745	<4.307	
113	-	NMe2	3-Methylbenzoyl	-	INIG	Chemistry 743		
116	o	NMe2	Chemistry 748	Et	Me	н	6.627	
			à				4 400	
117	0	CH2NMe2	3-Methylbenzyl	(CH2)4	(CH2)4	Н	<4.139	
118	3 o	NH2	3.5-Dimethylbenzyl	Ms	l-Pr	н	4.042	
111	90	NMs2	3.5-Dimethy/benzyl	Мв	i-Pr	н	6.114	
			3.				5 000	
12	00	NH2	3-Methoxybenzyl	Et	Me	Н	5.033	

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404	_			Eı	Me	н	8.469	6.948
121	0	NMe2	3-Methoxybenzyl	E1	MB	n	0.403	0.545
			ገ				1	
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122	0	NMe2	3-OHbenzyl	Et	Me	н	7.196	
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123	0	Chemistry 789	3,5-Dimethylbenzyl	Et	Me	н	8.444	6.918
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1								
				_			4.389	
124	0	NH2	Chemistry 796	Et	Ме	H	4.369	
			1 1					
	1						İ	
125	0	инсно	3-Methylbenzyl	Et	Ма	н		_
			×_°					
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126	0	инсно	3-Methylbenzoyl	Et	Me	н		
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127	,	NMe2	Chemistry 814	Et	Me	н	4.174	
121	۲	INME2	Cristiasty 014					
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	ļ		Но		}		İ	
128	30	NMe2	Chemistry 820	Et	Me	н	7.848	
			×	ł				
		**\\						
129	90	Chemistry 825	3,5-Dimethylbenzyl	Et	Me	H	8.398	7.057
		1	УОН	1				•
1								
12			Chamicias 922	 	Ме	н	<4	
13	00	NH2	Chemistry 832	Et	mo	<u> </u>		
			1 1			1		
				1				
13	10	NH2	3-Methylbenzyl	(CH2)3	(CH2)3	н	5.799	

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132	<u> </u>	NMe2	3-Methylbenzyl	(CH2)3	(CH2)3	н	7.863	
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			/ ^N -/					
						l., i		1
133	0	NMe2	Chemistry 850	Et	Me	H	4.94	
			ነ ጓ				1	
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134	٥	NH2	Chemistry 856	Et	Me	н	4.056	
			ጓ					
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135	0	NM82	Chemistry 862	Et	Me	н	6.688	
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		x ^N √₀∕]		1
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136	0		3-Methylbenzyl	Et	Мө	н	9	6.996
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137	s	NMe2	3,5-Dimethylbenzyl	Et	Me	н	7.658	
			×,0					}
		1			ļ	1		
	1							
138	s	NMe2	3.5-Dimethy/benzoyl	Eı	Мө	н	8.215	7.401
			×					
				1				
			\\\right\rac{1}{\right\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\rac{1}{\right\cin\ti\citi\citi\citi\citi\citi\citi\c		1			
139	0	инме	3-Trifluoramethylbenzyl	Et	Ме	н	6.908	
			V 40					
1			1			1		
1								
			\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\				1	
140		NH2	F 3-Trilluoromethylbenzoyl	Et	Мө	н	5.766	
1	1			1		† 		
			1 1					į
				1				1
144	10	NH2	Chemistry 898	Et	Мо	н	4.642	1

			X°					
142	0	NH2	3-Methylbenzoyl	(CH2)3	(CH2)3	н	4.889	
			×°°					
	j							
143	_	NMe2	Chemistry 910	Ει	Me	н	7.421	
.,,			ሻ					
		x ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~						
144	0	Chemistry 915	3-Methylbenzyl	Et	Ма	н	6.446	
		χ ^N ∕OH						
145	0	Chemistry 921	3-Methylbenzyl	Et	Me	н	8.42	6.028
		ı	**					
		O NH						
146		Chemistry 927	Chemistry 928	EL	Me	н		
140	_	Chemistry 327	ኝ					
		:						
147	0	NMe2	Chemistry 934	Et	Ме	н	7.721	
			× _p °.					
148	0	NMe2	3-Methylbenzoyl	(CH2)3	(CH2)3	н	7.863	
			×s					
149		NMe2	Chemistry 946	Et	Me	н	8.959	7,883
			×p°					
150		NH2	Chemistry 952	Et	Me	н	4.881	
			×°					
15	10	NMe2	Chemistry 958	Eı	Ма	н	7.845	

152 O NMe2 3.5-Dimetrybensyl Et Me Ph 4.21 153 O NMe2 3.5-Dimetrybensyl Et Me NPQ 6.749 154 O Chemistry 881 3-Methylaentyl Et Me N 8.009 6.262 155 O Chemistry 887 3-Methylaentyl Et Me N 9.334 155 O NH2 Chemistry 884 Et Me N 9.334 157 O NMe2 Chemistry 894 Et Me N 9.4334 158 O NMe2 Chemistry 1000 Et Me N 9.413 158 O NMe2 Chemistry 1000 Et Me N 9.413 158 O NMe2 Chemistry 1000 Et Me N 9.413	 -		·						
153 o NMe2 3.3-Dimetry/bencyl 51 Me NH2 6.749 154 o Chemistry 981 3-Methylbencyl 51 Me H 8.009 6.262 155 o Chemistry 987 3-Methylbencyl 51 Me H 7.514 156 o NH2 Chemistry 994 51 Me H 4.934 157 o NMe2 Chemistry 1000 51 Me H 6.413 158 o NMe2 Chemistry 1000 51 Me H 5.625 159 o NMe2 Chemistry 1012 51 Me H 7.011				ኣ					1
153 o NMe2 3.3-Dimetry/bencyl Et Me H 8.009 6.262 154 o Chemistry 987 3-Metry/bencyl Et Me H 7.514 155 o Chemistry 987 Chemistry 984 Et Me H 4.934 157 o NMe2 Chemistry 1000 Et Me H 6.413 158 o NMe2 Chemistry 1000 Et Me H 8.041 6.525 159 o NMe2 Chemistry 1012 Et Me H 7.011			•				1		j
153 o NMe2 3.3-Dimetry/bencyl Et Me H 8.009 6.262 154 o Chemistry 987 3-Metry/bencyl Et Me H 7.514 155 o Chemistry 987 Chemistry 984 Et Me H 4.934 157 o NMe2 Chemistry 1000 Et Me H 6.413 158 o NMe2 Chemistry 1000 Et Me H 8.041 6.525 159 o NMe2 Chemistry 1012 Et Me H 7.011									
154 o Chemistry 881 3-Methybenzyi Et Me H 8.009 6.262 155 o Chemistry 987 3-Methybenzyi Et Me H 7.514 156 o NH2 Chemistry 994 Et Me H 4.934 157 o NM62 Chemistry 1000 Et Me H 6.413 158 o NM62 Chemistry 1006 Et Me H 8.041 6.625 159 o NM2 Chemistry 1018 Et Me H 8.041 6.625	152	0	NMe2	3.5-Dimethylbenzyl	Et	Ме	Ph	4.21	
154 o Chemistry 981 3-Methylbentyl Et Me H 7.514 155 o Chemistry 987 3-Methylbentyl Et Me H 7.514 156 o NH2 Chemistry 1000 Et Me H 6.413 158 o NMa2 Chemistry 1006 Et Me H 8.041 6.625 159 o NH2 Chemistry 1018 Et Me H 7.011				ነ					
154 o Chemistry 981 3-Methylbentyl Et Me H 7.514 155 o Chemistry 987 3-Methylbentyl Et Me H 7.514 156 o NH2 Chemistry 1000 Et Me H 6.413 158 o NMa2 Chemistry 1006 Et Me H 8.041 6.625 159 o NH2 Chemistry 1018 Et Me H 7.011									
154 o Chemistry 981 3-Methylbentyl Et Me H 7.514 155 o Chemistry 987 3-Methylbentyl Et Me H 7.514 156 o NH2 Chemistry 1000 Et Me H 6.413 158 o NMa2 Chemistry 1006 Et Me H 8.041 6.625 159 o NH2 Chemistry 1018 Et Me H 7.011	150	_						. 740	
154 O Chemistry 981 3-Methylbensyl Et Me H 8.009 6.262 155 O Chemistry 987 3-Methylbensyl Et Me H 7.514 156 O NH2 Chemistry 894 Et Me H 4.934 157 O NMe2 Chemistry 1000 Et Me H 6.413 158 O NMe2 Chemistry 1000 Et Me H 8.041 6.625 159 O NH2 Chemistry 1012 Et Me H 7.011	153	0	NMe2	3,5-Uimethylbenzyl	Et	Ma	NH2	6.749	
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159 O NH2 Chemistry 1012 Et Ma H 7.011 160 O NMe2 Chemistry 1018 Et Ma H 8.678 7.177	}				}				
160 o NMe2 Chemistry 1018 Et Me H 8.678 7.177	158	0	NMe2	Chemistry 1006	Et	Me	н	8.041	6.625
160 o NMe2 Chemistry 1018 Et Me H 8.678 7.177		İ		⊁ _s			1		
160 o NMe2 Chemistry 1018 Et Me H 8.678 7.177					1		1		
160 o NMe2 Chemistry 1018 Et Me H 8.678 7.177									
160 o NMe2 Chemistry 1018 Et Me H 8.678 7.177	159	0	NH2		Et	Ме	H	7.011	
xh~~~	1			⊁ _{\$}		1			
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xh~~~	160							0 670	
161 O Chemistry 1023 3-Trilluoromethylbenzyl Et Me H 7.821 5.814	100	0	NMeZ	Chemistry 1018	E'	IVIE	ırı .	8,078	7.177
161 o Chemistry 1023 3-Trifluoromethylbenzyl Et Me H 7.821 5.814	1			۱ ٦					
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161 O Chemistry 1023 3-Trifluoromethylbenzyl Et Me H 7.821 5.814			A ~ 50	VF			1		İ
	161		Chemistry 1023	F 3-Trilluoromethylbenzyl	Et	Me	l _H	7.821	5.814
	1	Ť		\ \ \ \		1		=.	
	-] ,			1		
			1	TY	1		1		
162 O NMe2 Chemistry 1030 Et Me H 6.418 5.026	162	0	NMe2	Chemistry 1030	E	Мө	н	6.418	5.026

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4.00			Ž.				5 500	4.005
163	<u> </u>	NMe2	Chemistry 1036	Et	Me	н	5.596	4.236
164	0	XN Chemistry 1041	3-Methylbenzyl	Eì .	Ме	н	7.818	6.505
			× OH					
165	0	NMe2	Chemistry 1048	Et	Мв	н	4.354	<4
166		NMe2	Chemistry 1054	Et	Ме	н	5.693	4.518
· • • •	۳	-						
167	0.	NMe2	HN Chemistry 1060	Et	Me	н	6.338	5.828
	<u> </u>	, and a second						
168	o	NH2	Chemistry 1056	Et	Me	н	4.525	4.8 06
169	0	NMe2	Chemistry 1072	Et	Ме	н	7.101	5.771
170) o	NMe2	Chemistry 1078	Et_	Ме	н	8.553	7.224
			× × × × × × × × × × × × × × × × × × ×	5.			5.895	4.74
171	10	NMe2	Chemistry 1084	EI	Me	H	3.033	4.74
172	2 o_	NH2	3.5-Dimethylbenzyl	(CH2)4	(CH2)4	н	6.419	4.903
17	3 0	NMe2	3,5-Dimethylbenzyl	(CH2)4	(CH2)4	н	8.086	6.469
ننت								

			×°					
	i						ŀ	
174	0	NMe2	3-Bromobenzoyl	Et	Me	н	8.921	7.68
			×°					
		x ^N √_o′						
175	0	Chemistry 1107	3-Methylbenzoyl	Eı	Me	н	8.921	7.717
	<u> </u>	ordinastry vis	×NH					
							İ	
176	0	NMe2	Chemistry 1114	Et	Ме	н	8.432	6.436
			× ₆ °					
			s					
177	0	NH2	Chemistry 1120	Et	Ме	н	5.106	<4
			**°					
				İ				
178	0	NMe2	Chemistry 1126	EI	Me	H	7.873	6.461
179	0	NHMe	3-Bromobenzoyl	Et	Me	н	8.42	7.182
		XN O	1 1		ļ		l	
		x ^ñ √₀∕						
180	0	Chemistry 1137	3-Methylbenzyl	Et	Ме	н	5.988	
					Ì			
181	0	NMe2	Chemistry 1150	Et	Me	н	7.928	
			\ \mathcal{Y}^\circ}					
								,
			F				E 022	
182	6 0	NH2	Chemistry 1156	EI	Me	Н	5.933	
			F					
18	3 0	NMe2	Chemistry 1162	Et	Me	н	8.481	
	Ť		<u> </u>					
		x ^N ~~o′						
18	40	Chemistry 1187	3-Bramobenzyl	Et	Me	н	8.523	6.804
ئنا								

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		x*~	J.					
185	0	Chemistry 1173	3-Bromobenzoyl	Et	Me	н	8.745	7.433
400			, S				5704	
186	<u> </u>	NH2	Chemistry 1180	EI	Me	H	5.781	
187	0	NMe2	Br 186	Et	Ме	н	8.481	7.006
1			پ					
400			S Br				7.063	
188	<u> </u>	NH2	Chemistry 1192	Et	Me	H	7.063	
189	0	NH2	a,5-Dichlorobenzyl	Et	Мэ	н	6.401	
103	-	NHZ	5,5 Sichlorocality	<u> </u>	1 -			
190	0	NH2	c _I c _I	Et	Me	н	7.757	
191	0	NMe2	CI CI 3,5-Dichlorobenzyl	Et	Me	н	8.097	7.553
192	0	NMe2	3.5-Dichlorobenzoyl	Et .	Me	н	8.699	8.319
193		NMe2	Chemistry 1222	Et	Me	н	8.481	7.245
194		NH2	Chemistry 1228	Et	Me	н	4.665	
13.	₩-	INNE	Continuary 1220	+	 	- 	7.505	
19!	5 o	X ^N ✓ OH Chemistry 1233	3-Methylbenzyl	Et	Me	н	8.569	6.52

196		NMe2	Chemistry 1240	Et	Me		6.411	
	_	111102					1	
197	0	NH2	Chemistry 1246	El	Me	н	7.307	
	<u> </u>			<u> </u>				
198	0	NH2	Chemistry 1252	Ме	н	н	4.457	·
· · · · ·								
100	<u> </u>	x ^N ~~o		Et	Me	н	7.924	
199	<u> </u>	Chemistry 1257	3-Methylbenzyl	-	1410	<u> </u>		
200	0	Chemistry 1263	Benzyl	Et	Ме	н	8.42	5.95
			×	Et	Me	н	8.585	7.231
201	P	NMe2	Chemistry 1276	151	INIG	 		
202		NH2	2-Bromobenzyl	Et	Ме	н	5.715	
1-3-	 _							
203	0	NMe2	2-Bromobenzyl	Et	Me	н	8.161	

CLAIMS:

1. A compound having the formula (1)

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$$R^{6}$$
 $X-R^{7}$
 Q

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wherein:

- Q represents -NR₁R₂ or -R₀NR₁R₂ wherein:

*Ro represents C1-6 alkanediyl;

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* R₁ and R₂ each independently represent C₁₋₆alkyl or C₃₋₆alkenyl; said C₁₋₆alkyl and C₃₋₆alkenyl may be substituted with one, two or three substituents selected from hydroxy, C₁₋₄alkyloxy, C₁₋₄alkylthio, aryloxy, arylthio, amino, mono- or di(C1-4alkyl)amino and aryl; or

* R₁ and R₂ taken together may form a bivalent radical -R₁-R₂wherein - R_1 - R_2 - represents -(CH₂)₂-O-(CH₂)₂-, -(CH₂)₂-NR₇-(CH₂)₂,

 $-(CH_2)_2$ - $CH(NHR_7)$ - $(CH_2)_2$ - or $-(CH_2)_n$ wherein R_7 represents hydrogen or C₁₋₄alkyl and n represents 2, 3, 4, 5 or 6;

- R₃ represents aryl or a monocyclic or bicyclic heterocycle selected pyrimidinyl, thiazolinyl, furanyl, pyridinyl, benzoxazolyl, benzothiazolyl, benzimidazolyl; said monocyclic or bicyclic heterocycle may optionally be substituted with one, two or three substituents each independently selected from hydroxy, C1-4alkyl, C1-4alkoxy, halo,

trifluoromethyl, dimethylenoxy or phenyl,

-R₄ and R₅ each independently represent hydrogen, C₁₋₆alkyl,

thienyl,

imidazolyl,

 C_{3-6} alkenyl, C_{1-4} alkoxy, C_{1-4} alkyloxy C_{1-4} alkyl, amino, mono- or di(C_{1-4} alkyl) amino, formyl, C_{1-4} alkylcarbonyl carboxyl, C_{1-4} alkyloxycarbonyl, or C_{1-4} alkyl aminocarbonyl; wherein C_{1-6} alkyl and C_{3-6} alkenyl may be substituted with one, two or three substituents selected from hydroxy, C_{1-4} alkyloxy, C_{1-4} alkyl thio, aryloxy, arylthio, amino, mono- or di(C_{1-4} alkyl)amino and aryl; or R_4 and R_5 taken together form a bivalent radical of formula $-R_4-R_5$ -wherein $-R_4-R_5$ -represents -CH=CH-CH=CH- or $-(CH_2)_t$, wherein t represents -CH=CH-CH=CH- or $-(CH_2)_t$, wherein t represents -CH=CH-CH=CH-

- R₆represents hydrogen, hydroxy,C₁₋₄alkyloxy, c₁₋₆alkyl, C₃₋₆alkenyl, aryl, C₁₋₄alkyl, amino, mono- or di(C₁₋₄alkyl)amino or alkylaryl;
 - Y represents O or S;
 - X represents a radical of formula:

$$-(CH_2)_p$$
 (a) or $-(CH_2)_q$ - Z - $(CH_2)_r$ (b)

wherein p represents 1, 2, 3, 4 or 5;

q represents 0, 1, 2, 3, 4 or 5; r represents 0, 1, 2 or 3;

- Z represents NR8, C(= O), CHOH, CHNR8R9; CF2, O, S or CH=CH; wherein R8 and R9 each independently represent hydrogen or C1-4 alkyl;

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- a N-oxide, a stereochemically isomeric form or a pharmaceutically acceptable addition salt thereof.
- 2. A compound according to claim 1 wherein R_1 and R_2 represent each a methyl group.
- 3. A compound according to claim 1 wherein X represents -CH₂- and R₃ represents a phenyl group substituted with two methyl groups.
- 4. A compound according to claim 1 which is the 3-dimethylamino-4-(3,5-dimethylbenzyl)-5-ethyl-6-methylpyridin-2(1H)-one.

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- 5. A process for the obtention of compounds according to claim 1 wherein X represents - CH_{2} -, Y represents O, R_3 is an optionally phenyl group substituted and R_6 is hydrogen comprising the following steps:
- a) reacting a pyridine, substituted in position 2 with an alkoxy group and in position 3 with an amidoalkyl group, with a C_1 - C_6 alkyllithium, resulting in a lithiated derivate of the said pyridine.
- b) transforming said lithiated derivate into an organocopper reagent by reacting it with a complex formed by Cu I and dimethyl sulphide.
- c) obtaining a protected pyridinone by reacting the organocopper reagent with optionally substituted benzyl halide.
- d) hydrolysing said protected pyridinone and obtaining a deprotected pyridinone .
- e) substituting the amine-3 group of said deprotected pyridinone and obtaining the desired pyridinone compound.
- 6. A process for the obtention of compounds according to claim 1 wherein X represents -C(=0), Y represents O, R₃ is an optionally substituted phenyl group, and R₆ is hydrogen wherein:
- a) reacting a pyridine , substituted in position 2 with an alkoxy group and in position 3 with an amidoalkyl group, with a C_1 - C_6 alkyllithium, resulting in a lithiated derivate of said pyridine.
- b) reacting the lithiated derivative with an optionally substituted benzaldehyde, resulting in a substituted pyridinone,
- c) oxidizing said substituted pyridinone , resulting in a protected pyridinone ,
- d) deprotecting said protected pyridinone by hydrolysis, resulting in the desired pyridinone compound .
 - 7. Lithiated derivative having the following formula:

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wherein R_4 and R_5 are as defined in claim 1, and R_{10} and R_{11} are independently $C_1\text{-}C_6$ alkyl.

- 8. Pharmaceutical compositions comprising a therapeutically effective amount of at least a compound according to claim 1 and pharmaceutical carriers.
 - 9. Method of treatment of HIV-related diseases comprising the administration of an effective amount of a compound according to claim 1.
- 10. Method of treatment of HIV-infection comprising the administration of an effective amount of a compound according to claim 1.

INTERNATIONAL SEARCH REPORT

int errtional Application No PCT/EP 99/03023

A CLASSII IPC 6	FIGATION OF SUBJECT MATTER C07D213/73 A61K31/44 C07D417/06 C07D401/06	C07F1/02 C07D405/06	C07D213/74 C07D409/06	CO7D213/75		
According to	International Patent Classification (IPC) or to both no	ational classification an	d IPC			
	SEARCHED					
Minimum do IPC 6	cumentation searched (classification system follower CO7D A61K CO7F	d by classification symi	bols)			
Documentat	ion searched other than minimum documentation to t	he extent that such do	cuments are included in t	the fields searched		
Electronic d	ata base consulted during the international search (n	ame of data base and,	where practical, search	terms used)		
C. DOCUME	ENTS CONSIDERED TO BE RELEVANT			· · · · · · · · · · · · · · · · · · ·		
Category •	Citation of document, with indication, where approp	oriate, of the relevant p	assages	Relevant to claim No.		
X	WO 97 05113 A (CENTRE NAT; BISAGNI EMILE (FR); DOLL NG) 13 February 1997 (199 cited in the application claims; examples	.E VALERIE (1	T FR);	1-10		
Α	EP 0 462 800 A (MERCK & 0 27 December 1991 (1991-12 cited in the application claims			1-10		
Funt	her documents are listed in the continuation of box C	· X	Patent family member	s are listed in annex.		
"A" docume consid "E" earlier of filing d "L" docume which citation "O" docume other r "P" docume	later than the priority date claimed "&" document member of the same patent family					
Ì	actual completion of the International search	D	ate of mailing of the inter	national search report		
	7 July 1999		05/08/1999			
Name and r	nailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	A	uthorized officer Bosma, P			

..emational application No.

INTERNATIONAL SEARCH REPORT

PCT/EP 99/03023

Box I	Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)						
This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:							
1. X	Claims Nos.: 9 and 10 because they relate to subject matter not required to be searched by this Authority, namely: Remark: Although claims 9 and 10 are directed to a method of treatment of the human/animal body, the search has been carried out and based on the alleged effects of the compound/composition.						
2.	Claims Nos.: because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:						
3.	Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).						
Box II	Observations where unity of invention is lacking (Continuation of Item 2 of first sheet)						
This Inte	ernational Searching Authority found multiple inventions in this international application, as follows:						
1.	As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.						
2.	As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.						
3.	As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:						
4.	No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:						
Remark	The additional search fees were accompanied by the applicant's protest. No protest accompanied the payment of additional search fees.						

INTERNATIONAL SEARCH REPORT

information on patent family members

Int tional Application No PCT/EP 99/03023

	document earch report				atent family nember(s)		
WO 970	05113	A	13-02-1997	FR EP	2737496 A 0843663 A	07-02-1997 27-05-1998	
EP 04	52800	A	27-12-1991	AU AU CA FI JP JP JP	641769 B 7845291 A 2044828 A 912925 A 2079995 C 4253961 A 7107051 B 238576 A	30-09-1993 19-12-1991 19-12-1991 19-12-1991 09-08-1996 09-09-1992 15-11-1995 22-12-1994	
				PT US	98003 A 5308854 A	31-08-1993 03-05-1994 	

Form PCT/ISA/210 (patent family ennex) (July 1992)